## Montgomery County Community College CHE 261 Organic Chemistry I 4-3-3

COURSE DESCRIPTION:

| LE | ARNING OUTCOMES  | LEARNING ACTIVITIES   | EVALUATION METHODS                           |
|----|--|---|--|
| 4. | Identify the conformations of small alkanes and cycloalkanes, with emphasis on cyclohexane chair conformation with mono, di and tri substitutions. Draw Newman projections and predict stability of molecules. | Lectures Class Discussions Problem Solving Molecular Models Activity                                      | Section Exam and Comprehensive Final Exam    |
|    | Identify and draw radical and polar reaction mechanisms and use energy diagrams to identify transition states, intermediates and reactant-product energy changes.  | Lectures Class Discussions Problem Solving Emphasis on Reaction Mechanisms                                | Section Exam and Comprehensive Final Exam    |
| 6. | Explain properties of alkenes and discuss electrophilic addition with the use of carbocation rearrangement.  | Lectures Class Discussions Problem Solving Emphasis on Reaction Mechanisms Laboratory Activity and Report | Section Exam and Comprehensive Final Exam    |
| 7. | -  | Lectures Class Discussions Problem Solving Emphasis on Reaction Mechanisms Laboratory Activity and Report | Section Exam and<br>Comprehensive Final Exam |
| 8. | Name alkynes and discuss synthesis and reactions of alkynes: elimination, addition, hydration, reduction, oxidative cleavage and alkylation of acetylide amions.   | Lectures Class Discussions Problem Solving Emphasis on Reaction Mechanisms Laboratory Activity and Report | Section Exam and<br>Comprehensive Final Exam |

| LEARNING OUTCOMES   | LEARNING ACTIVITIES   | EVALUATION METHODS                           |
|---|---|--|
| 9. Identify enantiomers, with one or more chirality centers, R, S configuration, diastereomers, meso compounds and racemic mixtures.  | Lectures Class Discussions Problem Solving Molecular Models Activity                                      | Section Exam and<br>Comprehensive Final Exam |
| 10. Name alkyl halides and discuss synthesis: radical halogenation and allylic bromination and discuss reactions: use of Grignard Reagents and coupling Reactions.            | Lectures Class Discussions Problem Solving Laboratory Activity and Report Emphasis on Reaction Mechanisms | Section Exam and<br>Comprehensive Final Exam |
| 11. Discuss the mechanisms of alkyl halides: Sn1, SN2, E1, E1cb and E2; plus their kinetics and the   | Lectures Class Discussions Problem Solving Emphasis on Reaction Mechanisms Laboratory Activity and Report | Section Exam and<br>Comprehensive Final Exam |
| 12. Interpret mass spectra using fragmentation patterns and infrared spectra using location of common functional groups.  | Lectures Class Discussions Problem Solving Laboratory Activity and Report                                 | Section Exam and<br>Comprehensive Final Exam |
| 13. Interpret <sup>13</sup> C, DEPT <sup>13</sup> C and <sup>1</sup> H NMR using chemical shifts, integration and spinspin splitting techniques.                              | Lectures Class Discussions Problem Solving Laboratory Activity and Report                                 | Section Exam and<br>Comprehensive Final Exam |
| 14. Discuss properties of conjugated dienes: addition reaction, electrophilic addition, kinetic vs. thermodynamic control, Diels-Alder reaction and ultraviolet spectroscopy. | Lectures Class Discussions Problem Solving Emphasis on Reaction Mechanisms                                | Section Exam and Comprehensive Final Exam    |

At the conclusion of each semester/session, assessment of the learning outcomes will be completed by course faculty using the listed evaluation method(s). Aggregated results will be submitted to the Associate Vice President of Academic Affairs. The benchmark for each learning outcome is that 70% of students will meet or exceed outcome criteria.

## **SEQUENCE OF TOPICS:**

- I. Structure and Bonding
  - A. Atomic Structure: Orbitals and Electron Configuration
  - B. Valence Bond Theory
  - C. Hybridization: sp, sp<sup>2</sup> and sp<sup>3</sup> Orbitals
  - D. Molecular Orbital Theory
- II. Polar Covalent Bonds; Acids and Bases
  - A. Electronegativity
  - B. Dipole Moments
  - C. Formal Charges
  - D. Rules and Drawing Resonance Structures
  - E. Bronsted-Lowry Definition
  - F. Using pKa Values in Looking at Acid Strength
  - G. Organic Acids and Bases
  - H. Lewis Definition
  - I. Drawing Chemical Structures and Molecular Models
- III. Alkanes and Cycloalkanes
  - A. Functional Groups
  - B. Isomers
  - C. Alkyl Groups
  - D. Nomenclature
  - E. Properties
  - F. cis-trans Isomers of Cycloalkanes
- IV. Stereochemistry of Alkanes and Cycloalkanes
  - A. Conformations of Ethane, Propane and Butane
  - B. Baeyer Strain Theory
  - C. Conformations of Cyclopropane, Cyclobutane, Cyclopentane and Cyclohexane
  - D. Axial and Equatorial Bonds in Cyclohexane of Chair Conformation
  - E. Monosubstituted Cyclohexane
  - F. Conformational Analysis of di and tri Substituted Cyclohexane
  - G. Boat Cyclohexane
- V. Stereochemistry
  - A. Enantiomers
  - B. Chirality and Optical Activity
  - C. R, S Configuration
  - D. Diastereomers
  - E. Meso Compounds
  - F. Two Chirality Centers
  - G. Physical Properties of Stereoiosomers
  - H. Racemic Mixtures and their Resolution

- XI. Reactions of Alkyl Halides
  - Walden Inversion A.
  - Nucleophilic Substitution and Kinetics S<sub>N</sub>2 Reaction and Kinetics B.
  - C.
  - S<sub>N</sub>1 Reaction Kinetics and Stereochemistry D.
  - E. Eliminations ns

- 6. Dehydration of Cyclohexanol
  - a. Preparation of Cyclohexene Microscale
  - b. Test for Alkanes and Alkenes. with Bromine Water
  - c. Infrared Spectroscopy
- 7. Macroscale Addition of Bromine to Stilbene to Prepare Meso-Stilbene Dibromide
- 8. Microscale Nucleophilic Substitution Reactions
- 9. Infrared Spectroscopy; Nuclear Magnetic Resonance; Ultraviolet Spectroscopy and Mass Spectroscopy

## LEARNING MATERIALS:

McMurry, J. (2012). Organic Chemistry (8th ed.). Thomson and Brooks/Cole.

McMurry, S. (2012). *Study Guide and Student Solutions Manual* (7<sup>th</sup> ed.). Thomson and Brooks/Cole.

Williamson, K., Masters, K. (2011). *Macroscale and Microscale Organic Experiments* (6<sup>th</sup> ed.). Houghton-Mifflin Co.

Molecular Models

Other learning materials may be required and made available directly to the student and/or via the and/or course management system.

## **COURSE APPROVAL:**

| Prepared by: | Dr. E. Martins, Assistant F | Professor of Chemistry | Date: | 10/9/2004 |
|--------------|-----------------------------|------------------------|-------|-----------|
| Revised by:  | Dr. L. McAtee, Assistant F  | Professor of Chemistry | Date: | 2/5/2009  |
| VPAA/Provost | Compliance Verification:    | Dr. John C. Flynn, Jr. | Date: | 9/11/2009 |

Revised by: Dr. L. McAtee, Assistant Professor of Chemistry Date: 12/17/2012

VPAA/Provost or designee Compliance Verification:

Victoria L. Bastecki-Perez, Ed.D. Date: 2/13/2013

Revised by: Debbie Dalrymple Date: 6/27/2016

VPAA/Provost or designee Compliance Verification:

Victoria L. Bastecki-Perez, Ed.D. Date: 6/27/2016